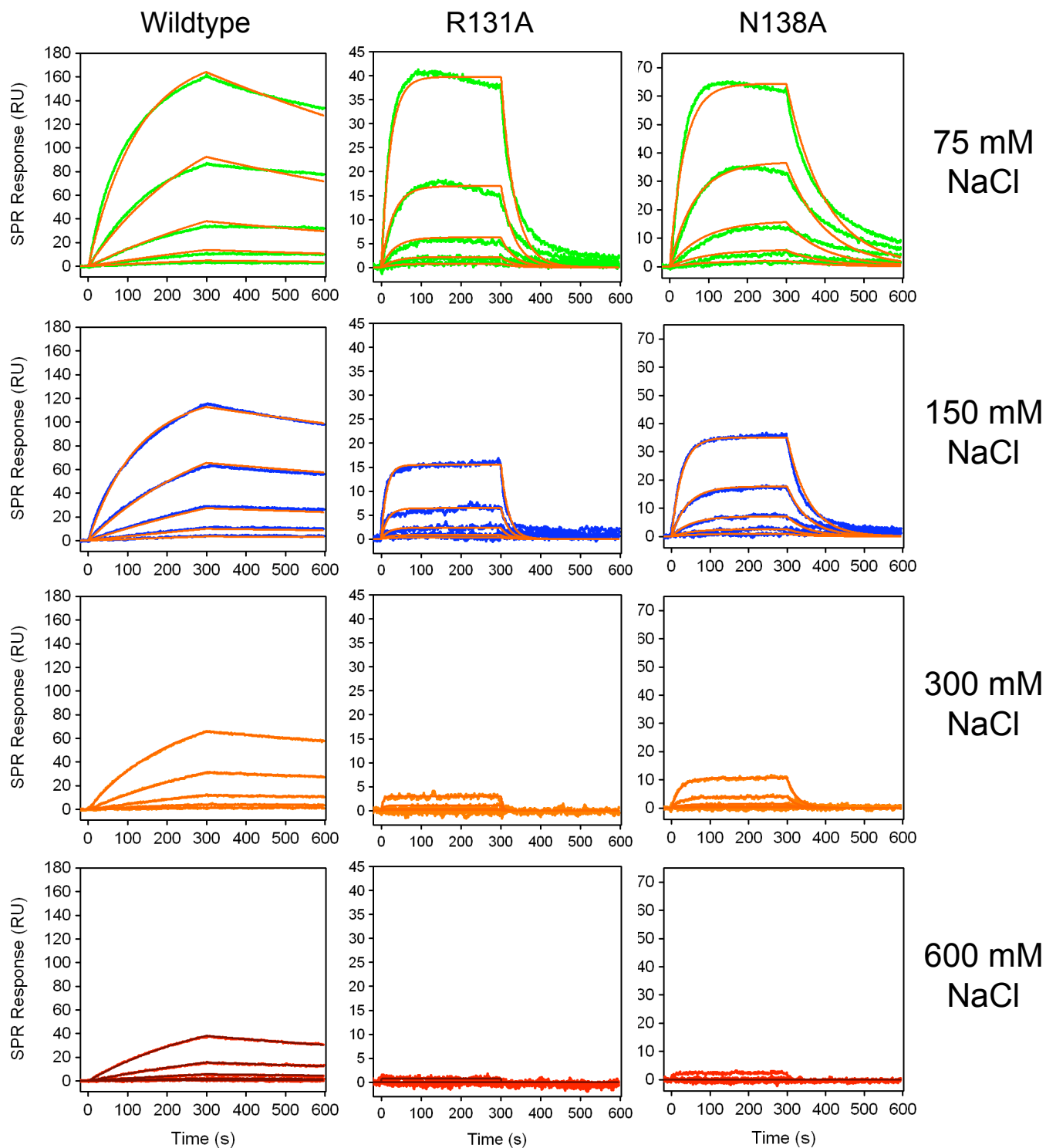
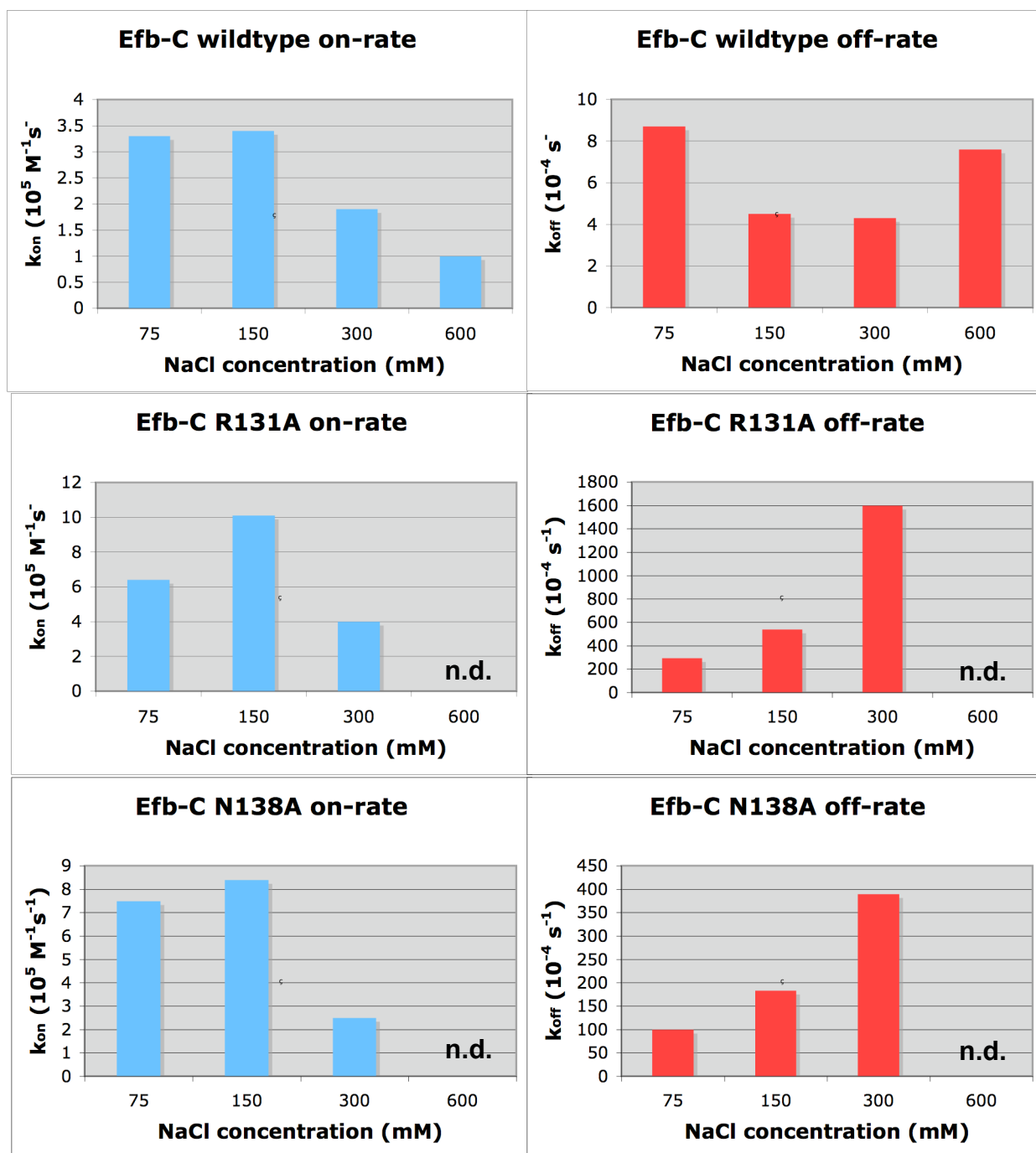


## Supplementary Figure 1



Kinetic evaluation of recombinant C3d (0.3-22 nM) interaction with immobilized Efb-C or its R131A and N138A mutant in PBS buffer of increasing salt concentration. All data sets were processed using Scrubber (v2; BioLogic) and fitted to a 1:1 Langmuir binding isotherm (thin red lines). Sub-physiological salt concentration (75 mM) lead to slight deviations from the 1:1 binding model, most likely due to increased non-specific binding. At high salt concentrations (300-600 mM), both mutants show very low signal-to noise ratios, which render a precise kinetic fit increasingly difficult. The resulting kinetic rate constants are plotted and compared in Supplemental Figure 2.

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**Supplementary Figure 2**



Comparison of the kinetic rate constants of the C3d interaction with Efb-C and its mutants R131A and N138A (based on the fitting results as seen in Supplemental Figure 1). In case of wildtype Efb-C, the effect of increasing buffer salt concentration is more pronounced on the kinetic on-rate (i.e. complex formation) than on the off-rate (complex stability). For the mutants, this trend is less clear and both rates seem to be affected. Note that the unfavorable signal-to-noise ratio, especially at high NaCl concentrations, make the kinetic evaluation increasingly imprecise (no rate constants could be extracted at 600 mM NaCl for both mutants). As a consequence, these results represent a semi-quantitative trend analysis rather than a quantitative assessment of kinetic profiles.

# Electrostatic Contributions Drive the Interaction Between *Staphylococcus aureus* Protein Efb-C and its Complement Target C3d

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## Supplementary Tables

Table 1: Energetic contribution of each residue on the wildtype Efb-C and two mutants to the complex binding. Notable contributions are marked in boldface. All the energies are expressed in kCal/mol.

Res. No.	WT	N138A	R131A	Res. No.	WT	N138A	R131A	Res. No.	WT	N138A	R131A
101	0.07	0.28	1.38	123	-0.13	-0.17	-0.14	145	-2.57	0.18	0.28
102	0.36	0.34	-1.88	124	0.10	0.06	0.07	146	-0.03	0.03	0.04
103	0.06	-1.00	-1.65	125	-0.71	0.02	0.04	147	0.10	0.12	0.13
104	-0.15	-1.34	-1.08	126	-4.30	-0.06	0.00	148	-0.28	-1.46	-1.85
105	-0.2	-0.97	-0.10	<b>127</b>	<b>-7.58</b>	<b>-4.03</b>	<b>-2.33</b>	149	-0.73	-0.04	-0.08
<b>106</b>	<b>-4.12</b>	<b>-3.88</b>	<b>-2.21</b>	128	0.13	-3.00	-0.38	150	0.11	0.10	0.09
107	-0.80	-1.55	-1.13	129	0.07	-0.02	0.01	151	0.10	0.09	0.11
108	0.36	0.32	0.31	<b>130</b>	<b>-1.63</b>	<b>-3.39</b>	<b>-3.12</b>	152	-0.62	0.24	-0.04
109	0.12	0.19	0.14	<b>131</b>	<b>-10.33</b>	<b>-11.03</b>	<b>-1.31</b>	153	0.31	0.20	0.21
<b>110</b>	<b>-3.16</b>	<b>-3.63</b>	<b>-3.02</b>	132	-0.12	-0.14	-0.60	154	0.06	0.05	0.09
111	-0.10	-0.06	-0.05	133	0.0	0.05	-0.05	155	-0.22	-0.05	-0.27
112	-0.01	-0.00	-0.00	<b>134</b>	<b>-2.41</b>	<b>-2.89</b>	<b>-2.79</b>	156	-1.35	-0.49	-0.80
113	0.14	0.13	0.06	<b>135</b>	<b>-3.56</b>	<b>-2.93</b>	<b>-4.49</b>	157	0.11	0.12	0.10
114	-0.03	-0.04	-0.02	136	0.01	0.00	-0.24	158	0.08	0.07	0.09
115	0.03	0.03	0.03	137	0.14	-0.98	-0.24	<b>159</b>	<b>-3.51</b>	<b>-2.82</b>	<b>-2.82</b>
116	0.03	0.07	0.05	<b>138</b>	<b>-2.81</b>	<b>-2.49</b>	<b>-5.75</b>	<b>160</b>	<b>-1.32</b>	<b>-1.52</b>	<b>-1.81</b>
117	0.04	0.04	0.04	<b>139</b>	<b>-3.83</b>	<b>-2.92</b>	<b>-3.69</b>	161	-0.26	0.05	0.01
118	0.06	0.10	0.08	140	-2.52	-0.84	-0.40	162	-2.76	0.01	-0.70
119	-0.16	-0.12	-0.11	<b>141</b>	<b>-1.25</b>	<b>-3.02</b>	<b>-3.24</b>	<b>163</b>	<b>-4.04</b>	<b>-2.54</b>	<b>-2.23</b>
120	0.33	0.57	0.39	<b>142</b>	<b>-5.16</b>	<b>-2.09</b>	<b>-2.82</b>	164	-0.10	-0.31	-0.15
121	-0.24	-0.17	-0.17	143	0.63	0.16	-0.20	<b>165</b>	<b>-10.53</b>	<b>-8.43</b>	<b>-8.25</b>
122	0.30	0.26	0.23	144	0.12	-0.04	-0.00				

Table 2: Energetic contribution of each residue on the wildtype C3d to the complex binding. Notable contributions are marked in boldface. All the energies are expressed in kCal/mol.

Res. No.	Energy	Res. No.	Energy	Res. No.	Energy
<b>39</b>	<b>-2.62</b>	53	-3.22	<b>106</b>	<b>-2.35</b>
<b>40</b>	<b>-2.62</b>	54	-0.04	<b>107</b>	<b>-2.73</b>
41	-0.01	55	0.10	<b>108</b>	<b>-2.03</b>
<b>42</b>	<b>-6.80</b>	56	-0.80	<b>166</b>	<b>-1.94</b>
43	0.09	57	-0.09	167	-0.89
44	-1.05	58	0.03	168	-0.53
<b>45</b>	<b>-2.07</b>	59	0.08	<b>169</b>	<b>-1.88</b>
46	0.63	60	-0.51	<b>170</b>	<b>-3.47</b>
47	0.01	<b>100</b>	<b>-1.65</b>	<b>171</b>	<b>-4.95</b>
48	0.10	<b>101</b>	<b>-1.76</b>	<b>172</b>	<b>-1.86</b>
<b>49</b>	<b>-2.32</b>	<b>102</b>	<b>-1.60</b>	173	-0.17
50	-1.32	<b>103</b>	<b>-1.89</b>	174	0.07
51	0.57	<b>104</b>	<b>-2.90</b>	175	0.00
<b>52</b>	<b>-2.78</b>	<b>105</b>	<b>-3.54</b>	176	-0.01